

Modern Chemistry Study Guide Section 5

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1. Match the equation type on the left to its representation on the right. synthesis(a)AX BY ? AY BX decomposition(b)A BX ? AX B single-displacement(c)A B ? AX double-displacement(d)AX ? A X. 2. In the reaction described by the equation $2Al(s) + 3Fe(NO_3)_3(aq) \rightarrow 2Al(NO_3)_3(aq) + 3Fe(s)$

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The formation of water according to the equation $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ shows that 2 molecules (made of 4 atoms) of hydrogen and 1 molecule (made of 2 atoms) of oxygen produce 2 molecules of water. The total mass of the product, water, is equal to the sum of the masses of each of the reactants, hydrogen and oxygen.

~~3 Atoms: The Building Blocks of Matter~~

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