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concentration is usually measured by the number of moles of substance dissolved in a liter of liquid. This is called molarity and is expressed as mol/L or M (The formula is: $M = \text{moles/Liter}$). Purpose: Practice molarity calculations in order to make 3 different solutions of Kool-Aid with the following concentrations: 0.1 M, 0.4 M, & 0.7M.

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Name: Per: Lab Activity- Kool-Aid Concentration

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what is the purpose of the lab? practice molarity calculations in order to make 3 different solutions of kool-aid with the following concentrations: 0.1 M, 0.4 M, 0.7 M and determine the concentration (molarity) of properly made kool-aid through a taste test.

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kool- aid concentration lab Flashcards | Quizlet

with Kool-Aid®, we will treat the name like a formula and give each letter the following fake molar masses: K = 2g O = .75g L = 3.5g A= 0.5g I = 1.5g D= 1g . CLAIM Solutions of identical concentration of solute should have the same absorbance of light, regardless of the method of preparation. PROCEDURE 1.

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Kool-Aid Molarity Lab. Introduction: This activity introduces you to solutions and allows you to experience making different concentrations of Kool- aid solution. There are many ways to calculate the concentration of a substance including: molarity (M), parts per million (ppm), percent composition (%)

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comp), and grams per liter (g/L). In chemistry, concentration is usually measured by the number of moles of substance dissolved in a liter of liquid.

Kool-Aid Molarity Lab - T.F. NORTH CHEMISTRY

Practice molarity calculations in order to make 3 different solutions of Kool-Aid with the following concentrations: 0.1 M, 0.4 M, & 0.7M . Determine the concentration (molarity) of properly made Kool-Aid through a taste test. Materials: Kool-Aid Powder, Popsicle sticks (to stir solutions), Water, Balance, 3 Plastic cups

Name Date Lab Activity- Kool-Aid Concentration

Absorbance at 640 nm versus Concentration of Red Dye #40

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(M) 0.8 0,7 0.6 0.5 0.4 0.3 y 1210.2x-0.0187 R20.9874 0 C0.2
0.1 0.00E+00 2.00E-04 4.00E-04 6.00E-04 8.00E-04

Concentration of Red Dye #40 (M) The absorbance of a Kool-Aid solution that had been diluted by a factor of three was 0.18.

Solved: In A Lab To Determine The Concentration Of Red Dye ...

Make sure to record all data and answer all the questions below. Step 1: You will prepare a 237.5 ml kool-aid solution with a concentration of 1M. The molar mass of kool-aid (or sugar since kool-aid is basically sugar $C_{12}H_{22}O_{11}$.) is 342 g/mol. Show the calculations below which helped. Subscribe to Unlock.

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Koolaid Molarity Lab Instructions.doc - Name Kool-aid ...

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Kool Aid Concentration Experiment Appropriate for grades 6-8 & Homeschool **As Featured in my 6th Grade Chemistry Unit Plan** PURPOSE: This experiment uses calculations and conversions to figure out how the concentration of a

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solution can be affected by changing either the amount of solute, solvent or both. These are vocabulary words that their high school teachers will surely thank you for ...

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