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that trades sodium ions for calcium and magnesium ions. After the column has been in use for some time, calcium and magnesium begin to appear in the water leaving the column.

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water softening: $2RNa^+ + Ca^{2+} \rightarrow R_2Ca^{2+} + 2Na^+$ The
exchanger R in the sodium

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ion form is able to exchange for calcium and thus, to remove calcium from hard water and replace it with an equivalent quantity of sodium. Subsequently, the calcium loaded resin may be treated with a sodium

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chloride solution,
regenerating it back to the
sodium form, so that it is
ready

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Softening The alkaline earths calcium, magnesium, strontium and barium dissolved in water are designated as water hardness. Strontium and barium play a subordinate and negligible

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role. With ion-exchange softening, the calcium and magnesium ions are exchanged for sodium ions. For this purpose, the water is passed over

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water by softeners. Their removal dramatically reduces the possibility of boiler scale formation. Water softening can be performed by ion exchange, whereby favorable sodium ions are

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exchanged for unfavorable
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hardness ions by the ion
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Eventually, the softener
resin bed exhausts (fills up
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part of the environment made and used by humans). The first chapter defines environmental chemistry and each of the five environmental spheres. The second chapter presents the basics of toxicological

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atmospheric quality. The author explains the nature of the geosphere and discusses soil for growing food as well as geosphere sustainability. He also describes the biosphere and its sustainability. The

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final sphere described is the anthrosphere. The text explains human influence on the environment, including climate, pollution in and by the anthrosphere, and means of sustaining this sphere. It also discusses renewable,

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through hydrometallurgical applications as diverse as the treatment of plating wastes and the tonnage production of uranium, to the present-day production of ultrapure water for the microelectronics industry,

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the recovery of valuable materials from sewage effluents and pollution control, the uses of ion exchange are legion. As a result, it is well-nigh impossible to prevent infiltration by the real

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world of even the most academic of conferences on the subject. It came as no surprise to the Scientific Board of the NATO Advanced Study Institute on "Mass Transfer & Kinetics of Ion Exchange" that one third of

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chemical periodicity. The text also highlights the electronic structures of atoms. Quantization of electricity; spectra of elements; quantization of the energy of an electron associated with nucleus; the

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