

# Chemical Reactions Pg 113

## Answer Key

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~~reactants and products  
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~~1.1 Deduce chemical equations when all  
reactants/products are given [SL IB~~

~~Chemistry]~~ **Identifying the reactants and  
products of an equation**

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~~Write the balanced equation for the~~

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following chemical reactions

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calcium w ate is commonly used in antacids

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reaction where one reactant yields two or  
more products. It is just an opposite of a  
synthesis or combination reaction.  
Displacement reaction Is a type of reaction  
in which a less reactive element is displaced  
by a more reactive element in a

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~~The products of this reaction are aqueous~~

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rusting nails examples of chemical change?  
they undergo one of these 4 chemical changes,  
formation of a gas, formation of a solid,  
changes in heat produced or absorbed &  
changes in color.

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