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$3e N V RT \times 18$

$\text{mol} \times 8.314 \times$

$10^{-5} (\text{bar} \cdot \text{m}^3$

$\text{mol K}) T P = =$

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$VV 1 \times 10^{-3} -$

$(10 - x) 106$

volume taken up  
by liquid

Equating  $P$  and  $P$

vap we have  $x 18$

$\times 8.314 \times 10^{-5} T$

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